Synthesis of Natural Atisane-Type Diterpenoids by *retro*-Biomimetic Transformations

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An efficient one step, *retro*-biomimetic procedure for the synthesis of natural products having the atisane structure is described (*Scheme 2*), natural products which are components of medicinal plants and possess relevant biological activity. Their structures were confirmed by chemical transformations and spectral data. The starting materials were the known *ent*-kaur-16-en-19-oic acid (1) and *ent*-trachyloban-19-oic acid (2), diterpenoids readily available from the waste of sunflower.

Introduction. – Tetracyclic atisanes, beyeranes, kauranes, and pentacyclic trachylobanes represent an important group of biosynthetically closely related polycyclic diterpenes, many of which display a wide range of biological activities [1-7]. According to the hypothesis of diterpene biogenesis [8], diterpenes belonging to the families of *ent*-beyerene (**A**), *ent*-kaurene (**B**), *ent*-trachylobane (**C**), and *ent*-atisene (**D**) might all arise from (–)-copalyl pyrophosphate **E** *via* nonclassical carbocations such as **F** as common intermediates (*Scheme 1*). This hypothesis [8] has been formulated on the basis of the known isoprene rule and, to the best of our knowledge, has not been turned down yet. *Scheme 1* provides a general overview of this biogenetic scheme, and our primary intention was to perform the *retro*-biomimetic transformation of *ent*-kaurene (**B**) to *ent*-beyerene (**A**) or *ent*-atisene (**D**) (path $\mathbf{B} \rightarrow \mathbf{F} \rightarrow \mathbf{A}$ or **D**).

The rearrangements of *ent*-kaurane- and *ent*-trachylobane-type diterpenes have been reported under the action of different reagents [9]. Most of the examples relate on the reactions involving the formation of the nonclassical carbocation of type **F** (*Scheme 1*). It is well-known from the work of *Olah* and co-workers [10] that superacids are very convenient generators of these species, and our own experience on the use of fluorosulfuric acid (= 'fluorosulfonic acid'; FSO₃H) as an efficient promoter of terpenoid cyclizations [11–15] and rearrangements [16–18] provided a motivation to investigate the behavior of *ent*-kaur-16-en-19-oic acid (1) [19] and *ent*-trachyloban-19-oic acid (2) under superacid treatment (*Scheme 2*).

Results and Discussion. – *ent*-Kaur-16-en-19-oic acid (1) and *ent*-trachyloban-19-oic acid (2) are accessible compounds and have been isolated from the dry waste of sun flower *Helianthus annuus* L. (*Scheme 2*) [20].

Treatment of *ent*-kaur-16-en-19-oic acid (1) with an excess (5 equiv.) of fluorosulfuric acid under mild reaction conditions (-60°) allowed carbonium ion generation and skeletal rearrangement to occur. The crude reaction product was first separated by column chromatography (CC) (SiO₂) in an unpolar-compounds fraction

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 $\textit{i)} \text{ FSO}_3\text{H} (5 \text{ equiv.}), \textit{i-PrNO}_2/\text{CH}_2\text{Cl}_2, \ -60^\circ, 15 \text{ min}, \text{ then } \text{Et}_3\text{N}. \textit{ii}) \text{ CH}_2\text{N}_2/\text{Et}_2\text{O}, \text{r.t.}, 30 \text{ min}.$

A and a polar-compounds fraction *B*. The unpolar compounds (*Fr. A*) were separated by another CC (silver nitrate-impregnated SiO₂, gradient elution) to afford, in the order of increasing polarity, *ent*-atis-15-en-19-oic acid (**3a**; 8%), *ent*-atis-16-en-19-oic acid (**4a**; 22%) [21], recovered starting material **1** (18%), *ent*-kaur-15-en-19-oic acid (**5a**; 17%), and finally *ent*-beyer-15-en-19-oic acid (**6a**; 7%) [21] (*Scheme 2*). The structures of the isolated products were established on the basis of spectral data; *i.e.*, their ¹H- and ¹³C-NMR spectra, assisted by DEPT and two-dimensional experiments (COSY, HSQC, HMBC) for signal assignment. Treatment of the individual acids **3a**, **4a**, **5a**, and **6a** with an Et₂O solution of diazomethane led to the corresponding esters, *i.e.*, methyl *ent*-atis-15-en-19-oate (**3b**) [22], methyl *ent*-atis-16-en-19-oate (**4b**) [21–24], methyl *ent*-kaur-15-en-19-oate (**5b**) [25], methyl *ent*-beyer-15-en-19-oate (**6b**) [21][26]. Their spectral properties matched the available literature data.

The polar compounds (*Fr. B*) were re-submitted to CC (SiO₂) to afford, in the order of increasing polarity; (16 α)-16-hydroxy-*ent*-atisan-19-oic acid (**7a**; 5%) [27] and (16 β)-16-hydroxy-*ent*-atisan-19-oic acid (**8a**; 4%) [27]. Their physicochemical and spectral properties matched the published data, and diazomethane methylation led to the corresponding methyl esters **7b** and **8b**, with physicochemical and spectral data matching those published earlier [28][29].

Thus, the superacid-promoted rearrangement of *ent*-kaurenoic acid **1** led predominantly to tetracyclic *ent*-atisane diterpenoids (see **3a**, **4a**, **7a**, and **8a**), with an overall yield of 39%. Taking into account the recovered starting material **1**, the combined yield of compounds **3a**, **4a**, **7a** and **8** amounted to *ca*. 62%.

Isomerization of *ent*-trachyloban-19-oic acid (2) was performed under conditions identical to those used for *ent*-kaur-16-en-19-oic acid (1). Superacid treatment afforded a mixture of rearranged compounds which were separated following the same protocol as described above for 1. From the unpolar-compounds fraction, *ent*-atis-15-en-19-oic acid (3a; 12%), *ent*-atis-16-en-19-oic acid (4a; 22%), *ent*-kaur-16-en-19-oic acid (1; 3%), *ent*-kaur-15-en-19-oic acid (5a; 1%), and finally *ent*-beyer-15-en-19-oic acid (6a; 2%) were obtained, identical by their spectral data with those formed from 1.

The polar-compounds fraction was separated into (16α) -16-hydroxy-*ent*-atisan-19oic acid (**7a**; 23%) and (16 β)-1b-hydroxy-*ent*-atisan-19-oic acid (**8a**; 16%), identical by their spectral data with those formed from **1**.

Thus, the superacid-promoted isomeritzation of *ent*-trachiloban-19-oic acid (2), yielded also essentially atisane diterpenoids, the overall yield of compounds 3a, 4a, 7a, and 8a amounting to *ca*. 73%.

The transformation of *ent*-kaur-16-en-19-oic acid (1) into *ent*-atisane-type compounds takes place by the formation of carbonium ion I, which rearranges *via* the nonclassical pentacyclic ion II to the *ent*-atisane carbonium ion III (*Scheme 3*). The latter undergoes a H-atom loss from either C(15) or C(17) to form the C=C bond isomeric *ent*-atisenoic acids **3a** and **4a**. The hydroxylated *ent*-atisanoic acids **7a** and **8a** are formed by quenching of III with a water molecule. On loss of one H-atom from C(15) of *ent*-kaurenoic acid carbonium ion I, the *ent*-kaur-15-en-19-oic acid (**5a**) is obtained. The *ent*-beyer-15-en-19-oic acid (**6a**) is formed after transformation of the nonclassical carbonium ion II to the *ent*-beyeranoic acid cation IV and subsequent H-atom loss from C(16).

Scheme 3. Proposed Reaction Course for Superacid-Promoted Isomerization of 1 and 2



Conclusions. – A new concept of natural-product synthesis is suggested, namely the *retro*-biomimetic approach. Based on this concept, easily accessible *ent*-kaurene-and *ent*-trachilobane-type diterpenoids were transformed to diterpenoids with an *ent*-atisane-type skeleton.

It is noteworthy mentioning that *ent*-atisane-type diterpenoids possess cytotoxic activity [30-33].

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Experimental Part

1. General. Treatment of reaction mixtures in org. solvents included the extraction by Et₂O, washing of the extract with H₂O up to neutral reaction, drying (Na₂SO₄), and solvent removal *in vacuo*. Column flash chromatography (CC): *Merck* silca gel 60 (70–230 mesh; *ASTM*). TLC: *Merck* precoated silica gel plates; detection by spraying with 0.1% cerium(IV) sulfate in 2N H₂SO₄ followed by heating at 80° for 5 min. M.p.: *Boethius* heating stage. Optical rotations: *Jasco-DIP-370* polarimeter; 5-cm cell; in CHCl₃. IR Spectra: *Spectrum-100* FT-IR spectrophotometer (*Perkin–Elmer*), with the universal ATR sampling accessory; $\tilde{\nu}$ in cm⁻¹. ¹H- and ¹³C-NMR Spectra: *Bruker-Avance-III* spectrometer (400.13 and 100.61 MHz); in CDCl₃; δ in ppm rel. to CHCl₃ as internal standard (δ (H) 7.26 and δ (C) 77.0), *J* in Hz. GC/MS: *Agilent-7890A* chromatograph; quadrupole MS detector *MSD 5975C*; *HP-5ms* capillary column (30 m/0.25 µm). EI-MS: operating at an ionization potential of 70 eV.; in *m/z* (rel. %), Elemental analyses: CHNOS analyzer *Vario EL III*.

2. Superacid-Promoted Isomerization of ent-Kaur-16-en-19-oic Acid (1). A soln. of 1 (403 mg, 1.324 mmol) in 2-nitropropane (2 ml) and CH₂Cl₂ (9 ml) was chilled to -60° and treated with a soln. of FSO₃H (662 mg, 6.62 mmol) in 2-nitropropane (1.4 ml), also chilled to -60° . After 15 min stirring at -60° , the mixture was quenched with Et₃N/hexane 1:1 (8 ml). Dilution with H₂O (10 ml) and usual workup gave a crude product (395 mg) which was submitted to CC (SiO₂ (10 g), AcOEt/petroleum ether 1:49, then 1:1): Fraction A of nonpolar compounds (298 mg, 74%) and Fraction B of polar compounds (74 mg, 18%). Fr. A (298 mg) was re-subjected to CC (silver nitrate impregnated SiO₂ (16 g), benzene and AcOEt/benzene): **3a** (33 mg, 8%; with benzene), **4a** (89 mg, 22%) and **1** (74 mg, 18%; with 1% AcOEt/benzene), and **5a** (68 mg, 17%) and **6a** (30 mg, 7%; with 3% AcOEt/benzene). Fr. B (74 mg) was

re-subjected to CC (SiO₂ (4 g), gradient AcOEt/benzene $7:93 \rightarrow 15:85$): **7a** (19 mg, 5%) and **8a** (16 mg, 4%).

3. Superacid-Promoted Isomerization of ent-Trachyloban-19-oic Acid (2). As described in Exper. 2, with 2 (297 mg, 0.938 mmol) in 2-nitropropane (1.6 ml) and CH₂Cl₂ (7.2 ml) and FSO₃H (692 mg, 6.92 mmol) in 2-nitropropane (1.2 ml). Quenching with Et₃N/hexane 1:1 (10 ml), dilution with H₂O (12 ml), usual workup, and CC (SiO₂ (7g)) of the crude product (290 mg) gave Fraction A of nonpolar compounds (123 mg, 41%) and Fraction B of polar compounds (150 mg, 51%). Fr. A (123 mg) was subjected to CC as described in Exper. 1: 3a (36 mg, 12%), 4a (66 mg, 22%) and 1 (8.3 mg, 3%), and 5a (3 mg, 1%) and 6a (5.5 mg, 2%). Fr. B (150 mg) was subjected to CC (SiO₂ (8 g)) as described in Exper. 1: 7a (68 mg, 23%) and 8a (47 mg, 16%).

ent-*Atis-15-en-19-oic Acid* (**3a**). White crystalline solid. M.p. 184–185° (MeOH). $[a]_{D}^{20} = -78.6 (c = 0.16, CHCl_3)$. IR (liquid film): 3550–3220, 3050, 1695, 1467, 1440, 1272, 1250. ¹H-NMR (400 MHz): 5.58 (br. *t*, *J* = 2, H–C(15)); 1.76 (*s*, Me(17)); 1.25 (*s*, Me(18)); 0.90 (*s*, Me(20)). ¹3C-NMR (100 MHz): 182.7 (*s*, C(19)); 140.3 (*s*, C(16)); 135.9 (*t*, C(15)); 57.2 (*d*, C(5)); 53.3 (*d*, C(9)); 43.7 (*s*, C(4)); 40.3 (*t*, C(1)); 38.0 (*t*, C(3)); 38.0 (*s*, C(10)); 37.9 (*t*, C(7)); 37.4 (*s*, C(8)); 35.9 (*d*, C(12)); 28.9 (*q*, C(18)); 28.3 (*t*, C(14)); 27.5 (*t*, C(11)); 26.6 (*t*, C(13)); 20.4 (*t*, C(6)); 20.0 (*q*, C(17)); 18.8 (*t*, C(2)); 11.9 (*q*, C(20)). EI-MS: 302 (12, M^+), 287 (10, $[M - Me]^+$), 274 (18), 256 (3), 241 (6), 207 (10), 159 (14), 119 (29), 106 (44), 81 (50), 44 (100). Anal. calc. for C₂₀H₃₀O₂: C 79.42, H 10.00; found: C 79.39, H 9.98.

ent-*Atis-16-en-19-oic Acid* (**4a**): White crystalline solid. M.p. $218-220^{\circ}$ (MeOH). ([21]: m.p. 219°) $[a]_{D}^{20} = -70.2 (c = 1.5, CHCl_3), ([21]: <math>[a]_D = -69.7 (c = 2.0, CHCl_3)$). IR (liquid film): $3550-3200, 3040, 2918, 1690, 1464, 1446, 1273, 1260. {}^{1}H-NMR (400 MHz): 4.73 (br.$ *d*,*J*= 2, 1 H); 4.57 (br.*d*,*J*= 2, 1 H); 2.22 (br.*s*, 1 H); 2.15 (br.*d*,*J*= 14, 1 H); 2.04 (br.*d*,*J*= 17, 1 H); 1.29 - 2.00 (*m*, 19 H); 1.25 (*s*, 3 H); 1.24 (*s*, 3 H); 0.91 - 1.18 (*m*, 8 H); 0.90 (*s*, 3 H); 0.70 - 0.88 (*m* $, 8 H). {}^{13}C-NMR (100 MHz): 182.3 ($ *s*, C(19)); 152.8 (*s*, C(15)); 104.5 (*t*, C(17)); 57.2 (*d*, C(5)); 52.2 (*d*, C(9)); 48.2 (*t*, C(16)); 43.7 (*s*, C(4)); 39.7 (*t*, C(1)); 39.7 (*t*, C(7)); 38.4 (*s*, C(10)); 38.1 (*t*, C(3)); 36.6 (*d*, C(12)); 33.6 (*s*, C(8)); 28.9 (*q*, C(18)); 28.7 (*t*, C(14)); 28.3 (*t*, C(13)); 27.3 (*t*, C(11)); 20.3 (*t*, C(6)); 18.8 (*t*, C(2)); 12.1 (*q*, C(20)). EI-MS: 302 (23,*M*⁺), 287 (75, [*M*- Me]⁺), 274 (4), 257 (21), 241 (27), 213 (15), 159 (21), 107 (58), 91 (100), 44 (86). Anal. calc. for C₂₀H₃₀O₂: C 79.42, H 10.00; found: C 79.37, H 9.96.

ent-*Kaur-15-en-19-oic Acid* (**5a**). White crystalline solid. M.p. 172–173° (MeOH). $[\alpha]_{D}^{20} = -59.2$ (c = 0.16, CHCl₃). IR (liquid film): 3540–3200, 3030, 1690, 1464, 1442, 1270, 1250. ¹H-NMR (400 MHz): 5.07 (br. t, J = 1.8, H–C(15)); 1.70 (s, Me(17)); 1.23 (s, Me(18)); 0.97 (s, Me(20)). ¹³C-NMR (100 MHz): 184.0 (s, C(19)); 142.5 (s, C(16)); 135.2 (t, C(15)); 56.9 (d, C(5)); 49.2 (s, C(8)); 48.2 (d, C(9)); 44.8 (d, C(13)); 43.9 (t, C(14)); 43.8 (s, C(4)); 40.8 (t, C(1)); 39.9 (s, C(10)); 39.5 (t, C(7)); 37.9 (t, C(3)); 28.9 (q, C(18)); 24.9 (t, C(12)); 20.8 (t, C(6)); 19.1 (t, C(11)); 19.0 (t, C(2)); 15.4 (q, C(20)); 15.4 (q, C(17)). EI-MS: 302 (13, M^+), 287 (9, [M - Me]⁺), 274 (3), 258 (5), 241 (8), 213 (5), 159 (16), 119 (34), 107 (62), 94 (100). Anal. calc. for $C_{20}H_{30}O_2$: C 79.42, H 10.00; found: C 79.41, H 9.97.

ent-*Beyer-15-en-19-oic Acid* (= (8 β ,13 β)-13-*Methyl-17-nor*-ent-*kaur-15-en-19-oic Acid*; **6a**): White crystalline solid. M.p. 183–185° (MeOH). ([21]: m.p. 184° (MeOH)) [a]_D²⁰ = +8.2 (c=0.33, CHCl₃), ([21]: [a]_D = +7.0 (c=1.8, CHCl₃)). IR (liquid film): 3025, 2935, 1685, 1445, 1255, 1190. ¹H-NMR (400 MHz): 5.76 (d, J = 5.7, H–C(15)); 5.47 (d, J = 5.7, H–C(16)); 1.26 (s, Me(18)); 1.01 (s, Me(17)); 0.69 (s, Me(20)). ¹³C-NMR (100 MHz): 184.3 (s, C(19)); 136.5 (d, C(16)); 134.8 (d, C(15)); 61.1 (t, C(14)); 57.2 (d, C(5)); 52.4 (d, C(9)); 49.2 (s, C(8)); 43.9 (s, C(4)); 43.7 (s, C(13)); 39.6 (t, C(1)); 38.0 (t, C(3)); 37.7 (t, C(7)); 37.7 (s, C(10)); 33.2 (t, C(12)); 29.1 (q, C(18)); 24.9 (q, C(17)); 21.6 (t, C(6)); 20.5 (t, C(11)); 19.3 (t, C(2)); 13.8 (q, C(20)). EI-MS: 302 (18, M^+), 287 (8, [M – Me]⁺, 272 (2), 257 (3), 207 (7), 148 (11), 135 (34), 105 (27), 91 (31), 44 (100). Anal. calc. for C₂₀H₃₀O₂: C 79.42, H 10.00; found: C 79.45, H 9.96.

(16β)-16-Hydroxy-ent-atisan-19-oic Acid (**7a**): White crystalline solid. M.p. 203–205° (MeOH) ([27]: m.p. 206°) $[a]_D^{10} = -31.2$ (c = 0.32, CHCl₃), [27]: $[a]_D = -10.81$ (c = 0.6, CHCl₃)). IR (liquid film): 3420, 2925, 1690, 1455, 1270. ¹H-NMR (400 MHz): 1.30 (s, Me(17)); 1.26 (s, Me(18)); 0.90 (s, Me(20)). ¹³C-NMR (100 MHz): 183.5 (s, C(19)); 72.4 (s, C(16)); 57.4 (t, C(15)); 57.2 (d, C(5)); 50.9 (d, C(9)); 43.7 (s, C(4)); 39.8 (t, C(7)); 39.8 (t, C(11)); 37.9 (d, C(12)); 38.2 (s, C(10)); 38.0 (t, C(3)); 33.7 (s, C(8)); 30.7 (q, C(17)); 28.9 (q, C(18)); 26.9 (t, C(14)); 25.5 (t, C(13)); 22.0 (t, C(11)); 20.0 (t, C(6)); 18.7 (t, C(2)); 12.1 (q, C(20)). EI-MS: 320 (2, M^+), 302 (40, [$M - H_2O$]⁺), 287 (51, [$M - H_2O$ -Me]⁺), 274 (13), 262 (57), 241 (23), 187 (32), 121 (68), 105 (85), 91 (100). Anal. calc. for $C_{20}H_{32}O_3$: C 74.96, H 10.06; found: C 74.89, H 10.02.

 $\begin{array}{l} (16a) - 16 - Hydroxy - \text{ent-}atisan - 19 - oic Acid (8a): White crystalline solid. M.p. 223 - 224° (MeOH) \\ ([27]: m.p. 225°) [a]_D^{10} = + 60.9 (c = 0.31, CHCl_3)) [27]: [a]_D = + 63.16 (c = 0.38, CHCl_3). IR (liquid film): 3350, 1690, 1460, 1440, 1270, 1260. ¹H-NMR (400 MHz): 1.31 (s, Me(17)); 1.26 (s, Me(18)); 0.90 (s, Me(20)). ¹³C-NMR (100 MHz): 183.0 (s, C(19)); 72.2 (s, C(16)); 57.4 (t, C(15)); 57.1 (d, C(5)); 50.6 (d, C(9)); 43.7 (s, C(4)); 39.7 (t, C(7)); 39.6 (t, C(1)); 38.0 (d, C(12)); 38.3 (s, C(10)); 38.0 (t, C(3)); 33.8 (s, C(8)); 30.5 (q, C(17)); 28.9 (q, C(18)); 26.8 (t, C(14)); 26.0 (t, C(11)); 23.9 (t, C(13)); 20.2 (t, C(6)); 18.7 (t, C(2)); 12.0 (q, C(20)). EI-MS: 302 (41, [M - H_2O]^+), 287 (100, [M - H_2O - Me]^+), 274 (7), 257 (24), 241 (21), 187 (22), 159 (27), 121 (41), 105 (50), 91 (54). Anal. calc. for C₂₀H₃₂O₃: C 74.96, H 10.06; found: C 74.91, H 10.07. \end{array}$

4. *Esterfications with Diazomethane: General Procedure (G.P.) Exemplified for Methyl* ent-*Atis-16-en-19-oate* (**4b**). Acid **4a** (10.5 mg) was treated with a sat. soln. of CH_2N_2 in Et_2O (1.0 ml). After 20 min, the solvent was evaporated and the residue purified by CC (SiO₂ (0.5 g), light petroleum ether): 9.1 mg of **4b**. White crystalline solid. M.p. 124–125° (MeOH) ([21]: m.p. 125–126° (MeOH); [22]: m.p. 126–127° (MeOH); [23]: m.p. 125–128° (MeOH); [24]: (\pm) **4b**: m.p. 102.5–105°). [α]₂₀²⁰ = –67.3 (c =0.35, CHCl₃) [21]: [α]_D = –68.7 (c =1.5, CHCl₃)). IR (liquid film): 3045, 1725, 1460, 1443, 1230. ¹H-NMR (400 MHz): 4.74 (br. d, J = 2, H_A–C(17)); 4.58 (br. d, J = 2, H_B–C(17)); 3.66 (s, MeO); 1.17 (s, Me(18)); 0.81 (s, Me(20)). ¹³C-NMR (100 MHz): 177.9 (s, C(19)); 152.8 (s, C(15)); 104.5 (t, C(17)); 57.3 (d, C(5)); 52.2 (d, C(9)); 51.0 (q, MeO); 48.2 (t, C(16)); 43.9 (s, C(4)); 39.7 (t, C(14)); 28.3 (t, C(13)); 27.3 (t, C(11)); 20.3 (t, C(6)); 18.8 (t, C(2)); 11.9 (q, C(20)).

Methyl ent-*Atis-15-en-19-oate* (**3b**): Accorfing to the *G.P.*, with **3a** (9 mg): 7.7 mg of **3b**. White crystalline solid. M.p. $89-90^{\circ}$ (MeOH) ([22]: m.p. $90-91^{\circ}$). $[a]_{D}^{20} = -81.6$ (c = 0.16, CHCl₃). IR (liquid film): 2920, 1690, 1465, 1443, 1275, 1258. ¹H-NMR (400 MHz): 5.58 (br. t, J = 1.8, H-C(15)); 3.64 (s, MeO); 1.73 (s, Me(17)); 1.17 (s, Me(18)); 0.77 (s, Me(20)). ¹³C-NMR (100 MHz): 178.1 (s, C(19)); 140.3 (s, C(16)); 135.8 (t, C(15)); 57.2 (d, C(5)); 53.3 (d, C(9)); 51.1 (q, MeO); 43.7 (s, C(4)); 40.3 (t, C(1)); 38.2 (t, C(3)); 37.9 (s, C(10)); 37.8 (t, C(7)); 37.3 (s, C(8)); 35.8 (d, C(12)); 28.8 (q, C(18)); 28.3 (t, C(14)); 27.5 (t, C(11)); 26.4 (t, C(13)); 20.4 (t, C(6)); 20.1 (q, C(17)); 18.7 (t, C(2)); 11.6 (q, C(20)).

Methyl ent-*Kaur-15-en-19-oate* (**5b**). According to the *G.P.*, with **5a** (5 mg): 4.3 mg of **5b**. White crystalline solid. M.p. 78–79.5° (MeOH) ([26]: m.p. 79–80°). $[a]_D^{20} = -61.6$ (c = 0.18, CHCl₃), [26]: $[a]_D = -54.0$ (c = 4.1, CHCl₃)). IR (liquid film): 2915, 1686, 1460, 1445, 1271, 1256. ¹H-NMR (400 MHz): 5.06 (br. t, J = 2, H-C(15)); 3.63 (s, MeO); 1.69 (s, Me(17)); 1.15 (s, Me(18)); 0.84 (s, Me(20)). ¹³C-NMR (100 MHz): 178.1 (s, C(19)); 142.5 (s, C(16)); 135.1 (t, C(15)); 56.8 (d, C(5)); 51.1 (q, MeO); 49.1 (s, C(8)); 48.0 (d, C(9)); 44.7 (d, C(13)); 43.8 (s, C(4)); 43.8 (t, C(14)); 40.8 (t, C(1)); 39.7 (t, C(7)); 39.6 (s, C(10)); 38.3 (t, C(3)); 28.7 (q, C(18)); 24.9 (t, C(12)); 20.9 (t, C(6)); 19.1 (t, C(11)); 18.9 (t, C(2)); 15.5 (q, C(20)), 15.4 (q, C(17)).

Methyl ent-*Beyer-15-en-19-oate* (= *Methyl* (8b,13b)-13-*Methyl*-17-*nor*-ent-*kaur-15-en-19-oate*; **6b**). According to the *G.P.*, with **6a** (7 mg): 6.4 mg of **6b**. White crystalline solid. M.p. 116–117° (MeOH) ([21]: m.p. 118° (MeOH)). $[\alpha]_D^{20} = +6.3 (c = 0.18, CHCl_3) ([21]: [\alpha]_D = +5.0 (c = 1.5, CHCl_3)). IR (liquid film): 3050, 3015, 2935, 1715, 1440, 1240, 1150. ¹H-NMR (400 MHz): 5.72 ($ *d*,*J*= 5.7, H–C(15)); 5.44 (*d*,*J*= 5.7, H–C(16)); 3.66 (*s*, MeO); 1.20 (*s*, Me(18)); 1.17 (*s*, Me(17)); 0.55 (*s*, Me(20)). ¹³C-NMR (100 MHz): 178.1 (*s*, C(19)); 134.5 (*d*, C(16)); 134.7 (*d*, C(15)); 61.1 (*t*, C(14)); 57.0 (*d*, C(5)); 52.2 (*d*, C(9)); 51.1 (*q*, MeO); 49.1 (*s*, C(8)); 43.6 (*s*, C(4)); 43.8 (*s*, C(13)); 39.5 (*t*, C(1)); 38.2 (*t*, C(3)); 37.6 (*t*, C(7)); 37.7 (*s*, C(10)); 32.9 (*t*, C(12)); 28.9 (*q*, C(18)); 24.9 (*q*, C(17)); 21.6 (*t*, C(6)); 20.4 (*t*, C(11)); 19.3 (*t*, C(2)); 13.6 (*q*, C(20)).

Methyl (16β)-16-Hydroxy-ent-*atisan-19-oate* (**7b**). According to the *G.P.*, with **7a** (10 mg): 9.3 mg of **7b**. White crystalline solid. M.p. 146–148° (hexane) ([28]: m.p. 148–150°; [29]: m.p. 149.5–151.5° (hexane)). $[\alpha]_D^{20} = -34.3 \ (c = 0.35, \text{ CHCl}_3)$. ([28]: $[\alpha]_D^{26} = -36.73 \ (c = 0.802, \text{ CHCl}_3)$). IR (liquid film): 3385, 2916, 1690, 1464, 1272, 1257. ¹H-NMR (400 MHz): 3.64 (*s*, MeO); 1.31 (*s*, Me(17)); 1.18 (*s*, Me(18)); 0.77 (*s*, Me(20)). ¹³C-NMR (100 MHz): 178.0 (*s*, C(19)); 72.3 (*s*, C(16)); 57.4 (*t*, C(15)); 57.2 (*d*, C(5)); 51.1 (*q*, MeO); 50.8 (*d*, C(9)); 43.8 (*s*, C(4)); 39.8 (*t*, C(7)); 39.8 (*t*, C(1)); 37.9 (*d*, C(12)); 38.2 (*s*, C(10)); 38.9 (*s*, C(12)

(t, C(3)); 33.7 (s, C(8)); 30.7 (q, C(17)); 28.7 (q, C(18)); 26.9 (t, C(14)); 25.4 (t, C(13)); 21.9 (t, C(11)); 20.0 (t, C(6)); 18.8 (t, C(2)); 11.9 (q, C(20)).

Methyl (16a)-16-Hydroxy-ent-Atisan 19-oate (**8b**). According to the *G.P.*, with **8a** (8 mg): 7.2 mg of **8b**. White crystalline solid. M.p. 135–138° (hexane) ([28]: m.p. 139–141°; [29]: m.p. 134.5–137° (hexane)). $[a]_D^{20} = -48.2$ (c = 0.36, CHCl₃) [28]: $[a]_D^{26} = -51.76$ (c = 0.701, CHCl₃)). IR (liquid film): 3388, 2918, 1687, 1465, 1446, 1271, 1255. ¹H-NMR (400 MHz): 3.64 (s, MeO); 1.28 (s, Me(17)); 1.16 (s, Me(18)); 0.77 (s, Me(20)). ¹³C-NMR (100 MHz): 178.0 (s, C(19)); 72.1 (s, C(16)); 57.5 (t, C(15)); 57.2 (d, C(5)); 51.1 (q, MeO); 50.6 (d, C(9)); 43.9 (s, C(4)); 39.7 (t, C(7)); 39.7 (t, C(11)); 38.2 (s, C(10)); 38.1 (d, C(12)); 38.1 (t, C(3)); 33.8 (s, C(8)); 30.5 (q, C(17)); 28.7 (q, C(18)); 26.8 (t, C(14)); 23.9 (t, C(13)); 23.4 (t, C(11)); 20.2 (t, C(6)); 18.8 (t, C(2)); 11.9 (q, C(20)).

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